



Styrene (CAS 100-42-5, molecular weight 104.14) is a volatile, monoaromatic hydrocarbon with the structural formula of $C_6H_5CH=CH_2$, vapour pressure of 880 Pa, Henry's law constant of $305.48 \text{ Pa}\cdot\text{m}^3\cdot\text{mol}^{-1}$, and a log octanol-water partition coefficient of 3.05. Synonyms for styrene include vinylbenzene, vinylbenzol, phenylethylene, styrolene, styrol, styrole, ethenylbenzene, cinnamene, cinnamenol and cinnamol (Government of Canada 1993). Pure styrene has a sweet aromatic odour at low concentrations (0.02 ppm) (Hoshika et al. 1993) and a disagreeable odour at higher concentrations (e.g., 100 ppm) (EPS 1984; Bond 1989). Styrene is only sparingly soluble in water at approximately $300\text{--}350 \text{ mg}\cdot\text{L}^{-1}$ (Mackay et al. 1993).

The dominant use of styrene is to make polymers for the manufacture of plastics, synthetic rubbers, and latexes such as polystyrene, acrylonitrile-butadiene-styrene (ABS), styrene-acrylonitrile (SAN), and styrene-butadiene (SB). These finished products are used in packaging material, disposable food and drink containers (molded expandable polystyrene, EPS), pipes (ABS), automobile instrument panel windows, clear housewear items (SAN), automobile tires (SB elastomer), paint (SB latexes and styrene-maleic anhydride), ion exchange resins for water treatment (styrene divinylbenzene resins), other plastic products, and fibrous glass products (Santodonato et al. 1980; USEPA 1992; Government of Canada 1993). Styrene is also an ingredient in floor waxes and polishes, paints, adhesives, putty, metal cleaners, autobody fillers, fibreglass boats, and varnishes (Howard 1989).

Styrene is produced in Canada at two plants in Ontario and one in Alberta (Government of Canada 1993). Canadian production of styrene in 1990 was reported to be 718 kt, of which approximately 490 kt were exported (CIS 1991). It is used in several industries across Canada (Ontario, Quebec, Alberta, British Columbia, and Nova Scotia) (Government of Canada 1993).

Styrene can be released into the environment during any stage of its manufacture, transport, disposal, or use. In addition to anthropogenic sources, styrene occurs naturally in the sap of styraceous trees, in bituminous-coal, and in shale-oil tars (RSC 1989), and is a natural by-product of the fungal and microbial metabolisms of a few species (Chen and Pepler 1956; Clifford et al. 1969).

In 1994, Environment Canada's National Pollutant Release Inventory (NPRI) recorded emissions of 1793 t of styrene (NPRI homepage: <http://www.ec.gc.ca/pdb/npri.html>).

Styrene concentrations are low in surface waters, generally below $1 \mu\text{g}\cdot\text{L}^{-1}$ (detection limit $0.2 \mu\text{g}\cdot\text{L}^{-1}$) (Government of Canada 1993), but can reach higher levels due to localized discharge events (e.g., $47.0 \mu\text{g}\cdot\text{L}^{-1}$) (Barton 1994).

Dissolved styrene will rapidly volatilize to the atmosphere (Mackay et al. 1993). Fu and Alexander (1992) estimated the half-life for styrene in lake water and distilled water as 1–3 h and 6–7 h, respectively. The shorter half-life in lake water was explained as significant aerobic biodegradation in addition to volatilization. The half-life of styrene in rivers has been estimated at 3 h (Howard et al. 1991), which is strongly affected by water mixing. In ponds (shallow water) and lakes (deep water), the half-life of styrene has been modeled to be 3 d and 13 d, respectively (USEPA 1984).

Styrene can also partition into animal tissue, however, the log BCFs of 0.83 and 1.13 determined from goldfish (Ogata et al. 1984) indicate that the bioconcentration of styrene in aquatic organisms is not likely to be significant. Although the K_{ow} indicates a moderate tendency for styrene to partition into fat, the low BCF is likely a result of the relatively rapid metabolism and excretion of styrene from the organism (USEPA 1992).

Styrene is known to cause tainting in fish tissue. The concentration in water impairing the flavour of yellow perch (*Perca flavescens*) was reported to range from 0.15 to $0.25 \text{ mg}\cdot\text{L}^{-1}$ (Persson 1984). A concentration of

Table 1. Water quality guidelines for styrene for the protection of aquatic life (Environment Canada 1998).

| Aquatic life | Guideline value ($\mu\text{g}\cdot\text{L}^{-1}$) |
|--------------|---|
| Freshwater | 72* |
| Marine | NRG† |

* Interim guideline.

† No recommended guideline.

0.037 mg·L⁻¹ was found to impart an odour to water in one older study (Rosen et al. 1963), while Persson (1984) found odours detectable at 0.11 mg·L⁻¹.

Water Quality Guideline Derivation

The interim Canadian water quality guideline for styrene for the protection of freshwater life was developed based on the 1991 protocol (CCME 1991). For more details, see the supporting document (Environment Canada 1998).

Freshwater Life

The interim guideline for styrene for the protection of freshwater life is 72 µg·L⁻¹.

Rainbow trout (*Oncorhynchus mykiss*) fry were the most sensitive fish species tested, with a 96-h LC₅₀ of 4.1 mg·L⁻¹ (Exxon Biomedical 1993) and a 96-h LC₅₀ of 2.5 mg·L⁻¹ (Qureshi et al. 1982). For fathead minnows (*Pimephales promelas*), 96-h LC₅₀s of 10 mg·L⁻¹ and 32 mg·L⁻¹ are reported by Machado (1995) and Mattson et al. (1976), respectively. For bluegill sunfish (*Lepomis macrochirus*), goldfish (*Carassius auratus*), and guppies (*Lebistes reticulatus*), the 96-h LC₅₀ values were 25, 65, and 75 mg·L⁻¹, respectively (Pickering and Henderson 1966).

The cladoceran *Daphnia magna* was the most sensitive of the invertebrates studied. The 48-h EC₅₀s (immobilization) are reported at 4.7 mg·L⁻¹ (Putt 1995a),

23 mg·L⁻¹ (LeBlanc 1980), and 59 mg·L⁻¹ (Qureshi et al. 1982). *Hyaella azteca* was less sensitive than *D. magna*, with a 96-h LC₅₀ of 9.5 mg·L⁻¹ (Putt 1995b). Erben and Pišl (1993) reported a 48-h LC₅₀ of 69 mg·L⁻¹ for the isopod *Asellus aquaticus* and 580 mg·L⁻¹ for the snail *Lymnaea stagnalis*.

The only plant used in acceptable toxicity studies was the green alga *Selenastrum capricornutum*. The 72-h EC₅₀ (inhibition of cell density, chronic) was found to be 1.4 mg·L⁻¹, and the 96-h EC₅₀ was 0.72 mg·L⁻¹ (Hoberg 1995). This is the most sensitive organism in the available data set, and the interim guideline value was calculated by multiplying the 96-h EC₅₀ of 0.72 mg·L⁻¹ by a safety factor of 0.1, yielding an interim guideline value of 72 µg·L⁻¹.

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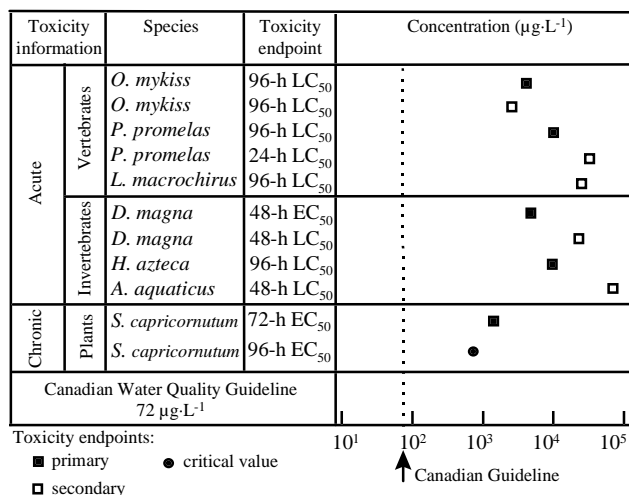


Figure 1. Select freshwater toxicity data for styrene.

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