



Canadian Water Quality Guidelines for the Protection of Aquatic Life

IPBC

IPBC (3-iodo-2-propynyl butylcarbamate (CAS number 55406-53-60 and molecular formula $C_8H_{12}INO_2$) is a carbamate compound (Konasewich and Quintin 1994) commonly used as a pesticide with various purposes, an industrial fungicide in paints and adhesives, an antimicrobial in metal-working fluids and on canvas and cordage, and as an antisapstain (Szenasy and Bailey 1996). It is a white crystalline powder with a specific odour. IPBC is highly soluble in organic solvents (e.g., acetone and benzyl alcohol) (Bioassay Systems Corporation 1982) and is moderately soluble in water ($156 \text{ mg}\cdot\text{L}^{-1}$ at 20°C) (Konasewich and St. Quintin 1994). IPBC has a low vapour pressure: $<0.002 \text{ Pa}$ at 20°C and 0.007 Pa at 30°C (USEPA 1997). The log octanol-water coefficient ($\log K_{ow}$) has been estimated at 2.81 (MRI 1990). IPBC is not likely to complex or absorb with suspended solids or sediments and is more likely to remain in the dissolved phase (Szenasy and Bailey 1996).

IPBC has received full registration in Canada for use as a material preservative in paints, adhesives, caulking, etc., and temporary full registration for use as an antisapstain wood preservative and joining wood preservative (M. Raphael 1998, Pest Management Regulatory Agency, Ottawa, pers. com.). IPBC was used in antisapstain formulations, primarily Kop-Coat NP-1 (7.6% IPBC), at 60% of mills surveyed in British Columbia in 1996 (Environment Canada 1998a). In 1996, lumber mills used 36 020 kg a.i. of IPBC for antisapstain purposes, reduced from the 47 540 kg used in 1993 (Environment Canada 1998a).

The mode of action of carbamate insecticides is primarily through acetylcholinesterase inhibition (Ecobichon 1991). The mode of action of IPBC, a fungicide and antimicrobial ingredient, however, is not clearly known, but may be linked to iodine toxicity (D. Nye 1998, Troy Corporation, Newark, New Jersey, pers. com.). The primary hydrolysis metabolite, propargyl butyl carbamate (PBC), has no iodine and is approximately 1000 times less toxic to fish and invertebrates than IPBC. Although iodine is an essential element, it is considered toxic at higher concentrations. Elemental iodine (I_2) is corrosive to exposed membranes and likely interferes with the permeability of the cell membranes and denatures proteins (Bowen, 1979; Santone and Prowis 1991).

The fate of IPBC in aquatic environments depends largely on physicochemical processes, particularly hydrolysis, rather than biotransformation processes. Microbial degradation is insignificant, due to the antimicrobial properties of IPBC (Schiefer 1990). EPL Bio-Analytical Services Inc. (1990a) reported IPBC to be hydrolytically stable at pH 5 with no signs of degradation, but at pH 7 the half-life was reported to be 139 d, and at pH 9, 0.947 d. Hydrolysis may be the main route of dissipation in aquatic environments, despite the dependence on alkalinity for catalyzation (USEPA 1997). Volatilization is not likely a significant dissipation route (Agriculture Canada et al. 1989), due to IPBC's low vapour pressure and its moderate solubility in water.

As IPBC is not known to be naturally occurring, all IPBC in the environment is expected to be from anthropogenic sources. Sources could include spills and other unpermitted discharges, permitted discharges from commercial facilities using the chemical, and discharges from products treated with IPBC (Henderson 1992). A sampling study initiated by the Fraser River Action Plan (FRAP) was abandoned following repeated sampling attempts at three separate mills using Kop-Coat NP-1, because IPBC concentrations in effluent samples were too low for accurate detection. The on-site dilution study performed used samples spiked with Kop-Coat NP-1 to determine dilution recoveries. IPBC recovery was unaffected by dilution with river water; recoveries reflected those anticipated by dilution calculations with distilled water. Although IPBC may be present at the effluent outfalls, dilution in sizable receiving waters would cause rapid dissipation (Szenasy 1998). Soil and water analysis near a leaking holding dike of a Kop-Coat NP-1 spill found no detectable IPBC where levels were expected to be $760 \text{ mg}\cdot\text{L}^{-1}$ (Koppers Company Inc. 1987).

Table 1. Water quality guidelines for IPBC for the protection of aquatic life (Environment Canada 1998b).

Aquatic life	Guideline value ($\mu\text{g}\cdot\text{L}^{-1}$)*
Freshwater	1.9*
Marine	NRG [†]

*Interim guideline.

[†]No recommended guideline.

Water Quality Guideline Derivation

The interim Canadian water quality guideline for IPBC for the protection of freshwater life was developed based on the CCME protocol (CCME 1991). For more information, see the supporting document (Environment Canada 1998b).

Freshwater Life

Acute toxicity data were found for six species of fish, ranging from a 96-h LC₅₀ of 0.067 mg a.i.·L⁻¹ for rainbow trout (*Oncorhynchus mykiss*) (Springborn Laboratories 1990) to a 96-h LC₅₀ of 1.9 mg P-100·L⁻¹ for coho salmon embryos (*O. kisutch*) (Farrell et al. 1998). Chronic toxicity data were found for only one fish species, the fathead minnow (*Pimephales promelas*). Embryos were exposed to technical grade IPBC in a flow-through system <24 h post-fertilization and were observed for survival at hatch (5 d) and for reduced larval weight and length at 35 d (Springborn Laboratories Inc. 1992). The 5-d LOEC for hatch survival was 0.057 mg·L⁻¹, while the 35-d LOEC for reduced weight gain and growth (length) was 0.019 mg·L⁻¹.

The toxicity of IPBC to invertebrates is represented by two species, *Daphnia magna* and *Hyalella azteca*, ranging from a 48-h LC₅₀ of 0.04 mg P-100·L⁻¹ (Farrell et al. 1998) to a 24-h LC₅₀ of 1.419 mg a.i.·L⁻¹ (Inveresk Research International 1989), both for *D. magna*.

The only plant study located was an exposure of *Chlorella pyreniodosa* in a static replacement system. IPBC was algistatic at 0.5 mg·L⁻¹ and algicidal at 1.0 mg·L⁻¹ (United States Testing Company Inc. 1988). This study, however, was ranked as unacceptable.

The lowest concentration causing a toxic effect was the 35-d LOEC of 0.019 mg a.i.·L⁻¹ for fathead minnows (Springborn Laboratories Inc. 1992). The guideline was derived by multiplying this LOEC by a safety factor of 0.1 (CCME 1991). This calculation results in an interim water quality guideline for IPBC of 1.9 µg·L⁻¹ (0.0019 mg·L⁻¹) for the protection of freshwater life

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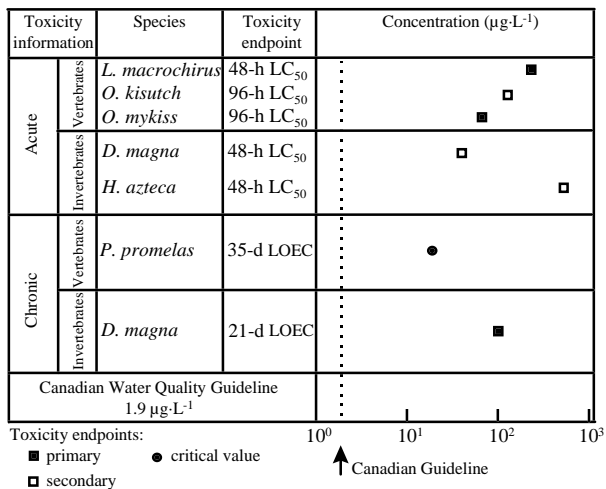


Figure 1. Select freshwater toxicity data for IPBC.

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For further scientific information, contact:

Environment Canada
Guidelines and Standards Division
351 St. Joseph Blvd.
Hull, QC K1A 0H3
Phone: (819) 953-1550
Facsimile: (819) 953-0461
E-mail: ceqg-rcqe@ec.gc.ca
Internet: <http://www.ec.gc.ca>

For additional copies, contact:

CCME Documents
c/o Manitoba Statutory Publications
200 Vaughan St.
Winnipeg, MB R3C 1T5
Phone: (204) 945-4664
Facsimile: (204) 945-7172
E-mail: spccme@chc.gov.mb.ca